(19) World Intellectual Property Organization International Bureau



(43) International Publication Date 3 June 2004 (03.05.2004)

PCT

## (10) International Publication Number WO 2004/046154 A1

(51) International Patent Classification<sup>3</sup>: 501/22

C07D 501/05,

(21) International Application Number:

PCT/IB2003/005032

(22) International Filing Date:

10 November 2003 (10.11.2003)

(25) Filing Language:

English

(26) Publication Language.

English

(30) Priority Date:

848/MAS/2002 15 No 152/MAS/2003 26 F

15 November 2002 (15.11.2002) IN 26 February 2003 (26.02.2003) IN

(71) Applicant (for all designated States except US): OR-CHID CHEMICALS & PHARMACEUTICALS LTD [IN/IN]; Orchid Towers., 313, Vallovar Kottam High Road, Nungambakkam, Chennai 600 034 (IN).

(72) Inventors; and

- (75) Inventors/Applicants (for US only): DESHPANDE, Pandurang, Bahwant [IN/IN]; C-1 "CEEBROS", Plot No. 32 (New) 1st Avenue, Indira nagar, Chemai 600 020 (IN). KHADANGALE, Bhausaheb, Pandharinath [IN/IN]; Santhi Avenue, No. 7, Dr. Radhakrishnan Road, Thisuvanmiyur, Chemai 600 042 (IN). RAMASUBBU, Chandraseharan [IN/IN]; No. 3 (New No.7) B1, Ramuiyam Poundation, 7th Main Road, Dhandeeswamagar, Velachery, Chenasi 600 042 (IN).
- (74) Common Representative: ORCHID CHEMICALS & PHARMACKUTICALS LTD; c/o Padmaja, S., Orchid Towers, 313, Valhavar Kottam High Road, Nungambakkam, Chennai 600034 (IN).
- (81) Designated States (nazional): AR, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DB, DK, DM, DZ, BC, EB, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR,

[Continued on next page]

#### (54) TIME: NOVEL AMORPHOUS HYDRATE OF A CEPHALOSPORIN ANTIBIOTIC

H<sub>2</sub>N H<sub>3</sub>S (2011)

(57) Abstract: A process for the preparation of coldinir of the formula (I) the said process comprising the steps of ii) condensing 7-amino-3-cephem-4-carboxylic acid of the formula (XII) wherein R1 is as defined above with compound of the formula (XIII) in the presence of a tertiary amine and an organic solvent, followed by treatment with a base to produce a sait of compound formula (XIV), wherein M+ is a counter ion and ii) hydrolyzing the compound of the formula (XIV) using an acid in the presence of a solvent to produce celdinir of formula (I).

#### WO 2004/046154 A1

KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SB, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.

(84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BR, BG, CH, CY, CZ, DR, DK, ER, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, ning of each regular issue of the PCT Gazette.

SE, SI, SK, TR), OAPI patent (BP, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NB, SN, TD, TG).

Declaration under Rule 4.17:

of inventorship (Rule 4.17(iv)) for US only

#### Published:

— with international search report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations\* appearing at the beginWO 2004/046154 PCT/IB2003/005032

## NOVEL AMORPHOUS HYDRATE OF A CEPHALOSPORIN ANTIBIOTIC

#### Field of the Invention

The present invention relates to a novel amorphous hydrate of a cephalosporin antibiotic. More particularly, the present invention relates to novel amorphous monohydrate of cefdinir of the formula (I).

5

15

20

The present invention also provides a process for the preparation of the novel amorphous monohydrate of cefdinir of formula (I).

The present invention also provides new salts of compound of formula (XIV) and a process for the preparation of cefdinir using the new salts.

#### Background of the Invention

Cefdinir is a third generation cephalosporin antibiotic for oral administration and has a broader antibacterial spectrum over the general gram positive and gram negative bacteria, especially against *Streptococci*, than other antibiotics for oral administration.

In view of the vital antibiotic activities of cefdinir of the formula (I), various methods of preparation were reported. Cefdinir is for the first time claimed in US patent No. 4,559,334 and the nature of the product that is disclosed in this patent is described as crystalline like amorphous in subsequent US patent (US 4,935,507). This patent also discloses a process for the preparation of cefdinir as depicted in the Scheme I.

CONFIRMATION COPY

10

15

Scheme I

In the disclosed process, 7-amino-3-vinyl-3-cephern-4-carboxylic acid ester where R represents a conventional carboxy protecting group, is acylated with the reactive ester of haloacylacetic acid, which was converted to its oxime, followed by cyclization with thiourea and deprotection of the ester group to afford cefdinir. The product obtained by the process described in examples 14 and 16 is approximately 80-85 % pure. The cyclization step suffers from poor yield and affords brownish color of the thiazole derivative, which subsequently affords cefdinir, but quality and yield were inferior. Further, owing to the fact that the expensive 7-amino-3-vinyl-3-cephern-4-carboxylic acid is carried through four steps, cost of producing cefdinir is high.

US patent number 4,935,507 claims the novel crystalline form of the cefdinir syn-isomer and a process for preparing the same. The X-ray crystallography data given in this patent is given in the following table:

2 θ ° Values	Relative Intensity
14.7	76
17.8	56
21.5	100
22.0	70
23.4	38

24.4	80
28.0	40

The crystalline form (Crystal A) of US 4,935,507 is prepared from the syn-isomer prepared according to the procedures described in Examples 14 and 16 of US 4,559,334.

5

10

In our US patent No. 6,388,070, we disclosed a process for preparing a compound of formula (VIII), wherein,  $R_1$  represents H, trityl, etc.,  $R_2$  represents H, phenyl, etc.,  $R_3$  represents CH<sub>3</sub>, CH=CH<sub>2</sub>, etc.,  $R_4$  is H or a salt or a carboxylic protecting group;  $R_5$  is H or trimethylsilyl; comprising acylating the compound of formula (VII) with compound of formula (VII) in the presence of an organic solvent, organic base and a silylating agent at a temperature in the range of -10 °C to +30 °C. The reaction is shown in scheme II below:

15

Scheme II

US patent No. 6,093,814 discloses a process for the preparation of cefdinir and its intermediate as represented in Scheme III:

WO 2004/046154 PCT/IB2003/005032

#### Scheme III

In this process p-methoxybenzyl 7-amino-3-vinyl-3-cephern-4-carboxylate condensed with 2-mercaptobenzothiazolyl (Z)-(2-amino-4-thiazolyl)-2-(trityloxyimino)acetate in N,N-dimethyl acetamide, and the product obtained was treated with p-toluenesulfonic acid in the presence of a mixture of diethyl ether and methanol to get crystalline 7-[(2-amino-4-thiazolyl)-2-(Z)-(trityloxyimino)acetamido]-3-vinyl-3-cephem-4-carboxylic acid.pTSA.2DMAc This process utilizes highly volatile, low-boiling and therefore industrially-not-preferred solvent, diethyl ether, for crystallizing out the above solvate. In addition, the quantity of the low-boiling solvent used is also very high ranging from 60-100 volumes, thereby adding hazard to the operations. Added to this is the fact that the recovery of these solvents from their mixture is not straight-forward.

15

10

US patent No. 6,350,869 discloses the purification of impure cefdinir through the preparation of N,N-dicyclohexylamine salt of 7-[2-amino-4-thiazolyl-2-(z)-hydroxyimino acetamido]-3-vinyl-3-cephem-4-carboxylic acid and

20

25

subsequent hydrolysis to get pure cefdinir. This process requires the preparation of crude cefdinir, conversion to N,N-dicyclohexylamine salt and then hydrolysis of the salt to get pure cefdinir, and therefore the overall yield is not attractive.

US patent No. 6,350,869 also emphasizes that cefdinir is unstable in the presence of other amines, with which, it gets heavily degraded. In addition, Yoshihiko Okamoto et al. (J. Pharm. Sci. Vol. 8(9), 976, 1996) report that cefdinir may be unstable under basic environment.

Crystalline cefdinir has limitations in formulation development as it cannot be developed into tablets.

Considering the foregoing limitations, we undertook an investigation in our lab to develop a product which is easy to handle and convenient to develop a dosage which is easily absorbable. We also parallel undertook an investigation to identify a process, which involves (i) less number of steps, (ii) the direct isolation of cefdinir, with out the need to prepare crude cefdinir in an additional step. This would permit commercializing the production of high-pure cefdinir with industrial-friendly solvent, which can further be recovered for recycling.

#### **Objectives of the Invention**

The main objective of the present invention is to provide a novel amorphous monohydrate of cefdinir which has very good bioavailability and useful in developing different dosage forms.

Another objective of the present invention is to provide a commercially viable process for the preparation of cefdinir and novel amorphous monohydrate of cefdinir of the formula (I), which would be easy to implement on manufacturing scale.

Yet another objective of the present invention is to provide new salts of formula (XIV), which are insoluble and stable throughout the process of

producing the cefdinir and a process for the preparation of cefdinir using these new salts.

## Summary of the Invention

In an embodiment of the present invention, there is provided process for the preparation of cefdinir of the formula (I)

H<sub>2</sub>N 
$$\stackrel{S}{\longrightarrow}$$
  $\stackrel{H}{\longrightarrow}$   $\stackrel{H}{\longrightarrow}$   $\stackrel{S}{\longrightarrow}$   $\stackrel{OH}{\longrightarrow}$   $\stackrel{O}{\longrightarrow}$   $\stackrel{O}{\longrightarrow}$ 

comprising the steps of:.

- i) condensing 7-amino-3-cephem-4-carboxylic acid of the formula (XII) wherein R<sub>1</sub> is as defined above with compound of the formula (XIII) in the presence of a tertiary amine and an organic solvent, followed by treatment with a base to produce a salt of compound formula (XIV), wherein M<sup>+</sup> is a counter ion and
- ii) hydrolyzing the compound of the formula (XIV) using an acid in the
   presence of a solvent to produce cefdinir of formula (I).

The reaction is shown in scheme-IV below:

## Scheme IV

Another embodiment of the present invention provides a novel amorphous monohydrate of cefdinir of the formula (I).

$$H_{N} \stackrel{S}{\longrightarrow} 0$$
 $H_{N} \stackrel{H}{\longrightarrow} H_{S}$ 
 $OH O COOH$ 
(1)

In yet another embodiment of the present invention, there is provided a process for the preparation of novel amorphous monohydrate of cefdinir of the formula (I) comprising hydrolyzing the compound of the formula (XV)

comprising the steps of:

10

i) adding an organic solvent to compound of formula (XV),

- ii) adjusting the pH of the resulting solution using an acid at a temperature in the range of 10 to 40 °C,
- iii) cooing the resulting solution rapidly to -40 to 0 and
- iv) isolating the novel amorphous monohydrate of cefdinir of the formula (I).

In yet another embodiment of the present invention, there is provided a process for the preparation of novel amorphous monohydrate of cefdinir of the formula (I) comprising hydrolyzing the compound of the formula (XV)

10 comprising the steps of:

- i) adding an organic solvent to compound of formula (XV),
- ii) cooing the resulting solution to -40 to 0 ° and
- iii) adjusting the pH of the resulting solution by rapid addition of an acid at a temperature in the range of 10 to 40 °C,
- 15 iv) isolating the novel amorphous monohydrate of cefdinir of the formula (I).

#### **Description of Figures**

Figure 1: Comparison of powder XRD pattern of the sample prepared according to US 4,935,507 and the sample prepared according to example 3 and example 4.

#### Detailed description of the invention

In an embodiment of the present invention, the activation group represented by X is selected from ester, thioester, halogen atom such as chlorine,

WO 2004/046154 PCT/IB2003/005032

as methyl, ethyl, n-propyl, iso-propyl, n-butyl or iso-butyl or a phenyl group; Alk group represents (C<sub>1</sub>-C<sub>4</sub>)alkyl group such as methyl, ethyl, n-propyl, iso-propyl, n-butyl or iso-butyl.

5

10

15

20

In an other embodiment of the present invention, the counter ion represented by M is selected from sodium, potassium, lithium, magnesium, ammonium, dicyclohexylamine, N,N'-dibenzylethylenediamine, 1,8-diazabicyclo(5.4.0)undec-7-ene (DBU), 1,5-diazabicyclo(4.3.0)non-5-ene, N,N'-diphenylethylenediamine, 1,4-dizabicyclo(2.2.2)octane, N,N-diisopropylethylamine, N,N-diisopropylamine and the like.

In another embodiment of the present invention, the tertiary amine used for condensation in step (i) is selected from triethylamine, N-methylpiperidine, N,N-diisopropylethylamine, trimethylamine and the like.

In yet another embodiment of the present invention, the organic solvent used for condensation in step (i) is selected from ethanol, methanol, isopropanol, THF, cyclohexanol, acetone, butan-2-one, acetonitrile, DMAc, water or a mixture thereof.

In yet another embodiment of the present invention, the base used for condensation in step (i) is selected from sodium hydroxide, sodium acetate, sodium 2-ethyl hexanoate, potassium hydroxide, ammonium hydroxide, ammonium acetate, calcium hydroxide, dicyclohexyl amine, N,N'-dibenzylethylenediamine diacetate, 1,8-diazabicyclo(5.4.0)undec-7-ene (DBU), 1,5-

10

15

20

diazabicyclo(4.3.0)non-5-ene, N,N'-diphenylethylenediamine, 1,4-diazabicyclo(2.2.2)octane, N,N-diisopropylethylamine, N,N-diisopropylamine, and the like.

In yet another embodiment of the present invention, the organic solvent used for hydrolysis is selected from acetone, 2-butanone, methanol, isopropanol, ethanol, THF, acetonitrile, DMAc, water and the like or mixtures thereof.

In another embodiment of the present invention, the hydrolysis is carried out using acid selected from HCl, sulfuric acid, formic acid, acetic acid, aromatic/aliphatic sulfonic acids such as benzenesulfonic acid, p-toluenesulfonic acid, naphthalenesulfonic acid, methanesulfonic acid, triflic acid, and the like.

In yet another embodiment of the present invention, the compound of formula (I) obtained is a syn isomer.

The present invention is based on the observation that rapid cooling of the aqueous solvent solution of cefdinir to low temperatures and adding the acid rapidly produces amorphous cefdinir. The technique can be achieved either by cooling the aqueous solvent solution to low temperatures and adding the acid rapidly to adjust the pH to precipitate the amorphous product or adding the acid to adjust the pH and rapidly cooling the resultant solution to precipitate the amorphous product.

In yet another embodiment of the present invention, there is provided new salts of compounds of formula (XIV)

wherein M<sup>+</sup> represents a counter ion as defined above.

The foregoing technique has been found to be markedly attractive, both from commercial point of view, as well as from manufacturing viewpoint and affords good quality of amorphous cefdinir of the formula (I).

Many other beneficial results can be obtained by applying disclosed invention in a different manner or by modifying the invention with the scope of disclosure.

The present invention is illustrated with the following examples, which should not be construed as limiting to the scope of the invention.

### Example 1

#### Step (1)

5

25

Preparation of 2-mercapto-S-phenyl-1,3,4-oxadiazolyl-(Z)-(2-aminothiazol-4-yl)-2-(trityloxyimino) acetate

To an ice-cold suspension of (Z)-(2-aminothioazol-4-yl)-2-(trityloxyimino)acetic acid (25 gm) in tetrahydrofuran (200 ml), triethylamine (10 gm) was added dropwise over 10 minutes at 0-5 °C. Bis-(2-oxo-oxazolidinyl)phosphinic chloride (15.4 gm) was added and stirred for one hour at 0-5 °C. To the reaction mixture 2-mercapto-5-phenyl-1,3,4-oxadiazole (9.8 gm) and triethylamine (5.0 gm) was added dropwise over 15 minutes and stirred at 0-5 °C for 6 – 7 hours. After completion of reaction, chilled water (500 ml) was added at 10-20 °C in 30 – 40 minutes and stirred at 20 °C for 2 hours. Then the slurry was cooled to 5-10 °C and stirred at this temperature for 45 minutes. The product thus obtained was filtered washed with water (100 ml) and dried at 30-35 °C for 4-5 hours to yield the title compound (50 gm, water content is 40%).

## Step (ii)

5

10

15

Preparation of potassium 7β-[2-(2-amino-4-thiazolyl)-2-(Z-trityloxylmino) acetamido]-3-vinyl-3-cephem-4-carboxylate

To a chilled suspension of 7-amino-3-vinyl-3-cephem-4-carboxylic acid (25 gm) and 2-mercapto-5-phenyl-1,3,4-oxadiazolyl-(Z)-(2-aminothiazol-4-yl)-2-(trityloxyimino)acetate (155 gm, water content is 40 %) in N,Ndimethylacetamide (150 ml), triethylamine (23 gm) was added drop-wise at 10±2 °C over 30-45 minutes and the resulting mixture was stirred at 20±2 °C for 6-8 hours. The reaction was monitored by HPLC. After completion of the reaction, tetrahydrofuran (125 ml), 10% sodium chloride solution (250 ml) and ethyl acetate (250 ml) were added at 25 °C and stirred for 20 min. The aqueous layer was separated and washed with ethyl acetate (250 ml). To the aqueous layer, ethyl acetate (500 ml) was added, cooled to 10 - 15 °C, and the pH was adjusted to 2.8-3.0 by 1:1 HCl in 30 min. The layers were separated and to the ethylacetate layer, 12 % (w/v) methanolic potassium hydroxide solution (60 ml) was added dropwise in 30 min at 25 °C, and stirred for 45 min. The resulting slurry was filtered, washed with ethyl acetate (150 ml) followed by acetone (150 ml) and dried at 30-35 °C under vacuum to obtain the title compound (45 gm, HPLC Purity >99.0%).

20

#### Step (iii)

Preparation of  $7\beta$ -[2-(2-amino-4-thiazolyl)-2-(Z-hydroxylmino)acetamido]-3-vinyl-3-cephem-4-carboxylic acid

A mixture of potassium 7β-[2-(2-amino-4-thiazolyl)-2-(Z-trityloxyimino)acetamido]-3-vinyl-3-cephem-4-carboxylate (25 gm) and activated carbon (2.5 gm) was added to an aqueous acetone solution (1:1, 70 ml) containing p-toluenesulphonic acid (17.7 gm) at 50 °C. The reaction mixture was heated to

70 °C in 20 minutes and maintained at this temperature for 35 minutes. After completion of the reaction, chilled ethylacetate (200 ml) having temperature -15 °C was added to the reaction mixture to reduce the temperature to 30-35 °C. The carbon was filtered and the carbon bed was washed with water (50 ml). The filtrate was diluted with water (200 ml), warmed to 35 °C and pH of the solution was adjusted to 6.0 -6.5 using aqueous ammonia solution (20%). The aqueous layer was separated and treated with carbon (2.0gm) at 35°C for 35 min. The carbon was filtered and the carbon bed was washed with water (50 ml). Acetone (25 ml) was added to the filtrate and 10 % (w/v) solution of sulphuric acid was added dropwise to bring down the pH from 4.5 to 2.8 at 33-35 °C, stirred for 30 minutes and adjusted the pH again to 2.6. The resulting slurry was stirred for 15 - 20 minutes at 33-35 °C, cooled to 20-25°C, and stirred for 30 minutes. The product thus obtained was filtered, washed with water (50 ml) and dried at 35 °C under vacuum for 3-4 hours to get the title compound (9.0 gm, HPLC purity < 99%).

## Example 2

#### Step (i)

10

15

# Preparation of 2-mercapto-5-phenyl-1,3,4-oxadiazolyl-(Z)-(2-aminothiazol-4-yl)-2-(trityloxyimino) acetate

To an ice-cold suspension of (Z)-(2-aminothioazol-4-yl)-2-(trityloxyimino)acetic acid (25 gm) in tetrahydrofuran (200 ml), triethylamine (10 gm) was added dropwise over 10 minutes at 0-5 °C. Bis-(2-oxo-oxazolidinyl)phosphinic chloride (15.4 gm) was added and stirred for one hour at 0-5 °C. To the reaction mixture 2-mercapto-5-phenyl-1,3,4-oxadiazole (9.8 gm) and triethylamine (5.0 gm) was added dropwise over 15 minutes and stirred at 0-5 °C for 6 - 7 hours. After

completion of reaction, chilled water (500 ml) was added at 10-20 °C in 30 – 40 minutes and stirred at 20 °C for 2 hours. Then the slurry was cooled to 5-10 °C and stirred at this temperature for 45 minutes. The product thus obtained was filtered washed with water (100 ml) and dried at 30-35 °C for 4-5 hours to yield the title compound (50 gm, water content is 40%).

#### Step (II)

Preparation of potassium 7β-[2-(2-amino-4-thiazolyl)-2-(Z-trityloxylmino) acetamidol-3-vinyl-3-cephem-4-carboxylate

To a chilled suspension of 7-amino-3-vinyl-3-cephem-4-carboxylic acid (5 gm) and 2-mercapto-5-phenyl-1,3,4-oxadiazolyl-(Z)-(2-aminothiazol-4-yl)-2-(trityloxyimino)acetate (24.2 gm) in tetrahydrofuran (40 ml) and water (5 ml), triethylamine (4.6 gm) was added drop-wise at 20±2 °C over 10-15 minutes and the resulting mixture was stirred at 30±2 °C for 6-8 hours. The progress of the reaction was monitored by HPLC. After completion of reaction, ethylacetate (100 ml) and water (75 ml) were added at 30±2 °C and stirred for 20 min. The aqueous layer was separated and washed with ethyl acetate (75 ml). To the aqueous layer, ethylacetate (150 ml) was added, cooled to 10 - 15 °C, and the pH was adjusted to 2.8-3.0 by 1:1 HCl solution in 25-30 min. To the separated ethylacetate layer, acetone (50 ml) and a methanolic potassium hydroxide solution (7.5 % w/v, 20 ml) were added dropwise in 25-30 min at 25 -27 °C and stirred for further 45 min. The resulting slurry was filtered, washed with acetone (2 X 25 ml) and dried at 30-35 °C under vacuum to obtain the title compound (5.0 gm, HPLC Purity >99.0 %).

20

#### Step (iii)

Preparation of 7β-[2-(2-amino-4-thiazolyl)-2-(Z-hydroxyimino)acetamido]-3-vinyl-3-cephem-4-carboxylic acld

 $7\beta$ -[2-(2-amino-4-thiazolyl)-2-(Zof potassium A mixture trityloxyimino)acetamido]-3-vinyl-3-cephem-4-carboxylate (25 gm) and activated carbon (2.5 gm) was added to an aqueous acetone solution (1:1, 70 ml) containing p-toluenesulphonic acid (17.7 gm) at 50 °C. The reaction mixture was heated to 70 °C in 20 minutes and maintained at this temperature for 35 minutes. After completion of the reaction, chilled ethylacetate (200 ml) having temperature -15 °C was added to the reaction mixture to reduce the temperature to 30-35 °C. The carbon was filtered and the carbon bed was washed with water (50 ml). The filtrate was diluted with water (200 ml), warmed to 35 °C and pH of the solution was adjusted to 6.0 -6.5 using aqueous ammonia solution (20%). The aqueous layer was separated and treated with carbon (2.0gm) at 35°C for 35 min. The carbon was filtered and the carbon bed was washed with water (50 ml). Acetone (25 ml) was added to the filtrate and 10 % (w/v) solution of sulphuric acid was added dropwise to bring down the pH from 4.5 to 2.8 at 33-35 °C, stirred for 30 minutes and adjusted the pH again to 2.6. The resulting slurry was stirred for 15 -20 minutes at 33-35 °C, cooled to 20-25°C, and stirred for 30 minutes. The crystals thus obtained was filtered, washed with water (50 ml) and dried at 35 °C under vacuum for 3-4 hours to get the title compound (9.0 gm, HPLC purity < 99%).

#### Example 3

15

20

Preparation of (Z)-7β-[2-(2-amino-4-thiazolyl)-2-(hydroxyimino)acetamido]3-vinyl-3-cephem-4-carboxylic acid in amorphous hydrate form

Ammonium (Z)-7β-[2-(2-amino-4-thiazolyl)-2-(hydroxyimino)acetamido]-3-vinyl-3-cephem-4-carboxylate (20 gm) was added to a mixture of water (250 ml) and acetone (80 ml) and warmed to 33-35 °C. This aqueous solution was treated with activated charcoal and EDTA at 35 °C for 40 minutes. The carbon was filtered and the carbon bed was washed with water (70 ml). This aqueous acetone solution was cooled to -30 °C and a (10 %) solution of aqueous sulphuric acid was added rapidly, stirred for 30 minutes and warmed to 0-2 °C. The product thus obtained was filtered at 0-2 °C, washed with cold-water (100 ml) and dried at 40-45 °C under vacuum for 5-6 hours to get (Z)-7β-[2-(2-amino-4-thiazolyl)-2-(hydroxyimino)acetamido]-3-vinyl-3-cephem-4-carboxylic acid (6.0 gm, HPLC quality 89.0 %, water content 4-5 %).

#### Example 4

10

20

25

Preparation of (Z)-7β-[2-(2-amino-4-thiazolyl)-2-(hydroxylmino)acetamido]-3-vinyl-3-cephem-4-carboxylic acid in amorphous hydrate form

Ammonium (Z)-7β-[2-(2-amino-4-thiazolyl)-2-(hydroxyimino)acetamido]-3-vinyl-3-cephem-4-carboxylate (20 gm) was added to a mixture of water (250 ml) and acetone (80 ml) and warmed to 33-35 °C. This aqueous solution was treated with activated charcoal and EDTA at 35 °C for 40 minutes. The carbon was filtered and the carbon bed was washed with water (70 ml). The pH of this aqueous acetone solution was adjusted to 0.6 at 33-35 °C using a (10 %) solution of aqueous sulphuric acid. This solution was cooled rapidly to -10 °C and stirred for 30 minutes. The product thus obtained was filtered at -10 °C, washed with cold-water (100 ml) and dried at 40-45 °C under vacuum for 5-6 hours to get (Z)-7β-[2-(2-amino-4-thiazolyl)-2-(hydroxyimino)acetamido]-3-vinyl-3-cephem-4-carboxylic acid (6.0 gm, HPLC quality 93.0 %, water content 4-5 %).

Claims:

1. A process for the preparation of cefdinir of the formula (1)

the said process comprising the steps of:

5 i) condensing 7-amino-3-cephem-4-carboxylic acid of the formula (XII)

wherein R<sub>1</sub> is as defined above with compound of the formula (XIII)

in the presence of a tertiary amine and an organic solvent, followed by treatment
with a base to produce a salt of compound formula (XIV),

wherein M<sup>+</sup> is a counter ion and

15

ii) hydrolyzing the compound of the formula (XIV) using an acid in the presence of a solvent to produce cefdinir of formula (I).

2. The process as claimed in claim 1, wherein activation group represented by X is selected from ester, thioester, halogen atom such as chlorine, bromine,

- where R<sub>6</sub> represents (C<sub>1</sub>-C<sub>4</sub>)alkyl group or phenyl group; Alk group represents (C<sub>1</sub>-C<sub>4</sub>)alkyl.
- The process as claimed in claim 1, wherein the counter ion represented by M is selected from sodium, potassium, lithium, magnesium, ammonium, dicyclohexylamine, N,N'-dibenzylethylenediamine, 1,8-diazabicyclo(5.4.0)undec-7-ene (DBU), 1,5-diazabicyclo(4.3.0)non-5-ene, N,N'-diphenylethylenediamine, 1,4-dizabicyclo(2.2.2)octane, N,N-diisopropylethylamine or N,N-diisopropylamine.
  - 4. The process as claimed in claim I, wherein the tertiary amine is selected from triethylamine, N-methylpiperidine, N,N-diisopropylethylamine, trimethylamine and the like.
- 5. The process as claimed in claim 1, wherein the organic solvent used in step(i) is selected from ethanol, methanol, isopropanol, THF, cyclohexanol, acetone, butan-2-one, acetonitrile, DMAc, water or a mixture thereof.
  - 6. The process as claimed in claim 1, wherein the organic solvent used in step(ii) is selected from acetone, 2-butanone, methanol, isopropanol, ethanol, THF, acetonitrile, DMAc, water and the like or mixtures thereof.
  - 7. The process as claimed in claim 1, wherein the acid is selected from HCl, sulfuric acid, formic acid, acetic acid or aromatic/aliphatic sulfonic acids.

- 8. The process as claimed in claim 1, wherein the compound of formula (1) obtained is a syn isomer.
- 9. A novel amorphous monohydrate of cefdinir of the formula (I)

10. The process for the preparation of novel amorphous monohydrate of cefdinir of the formula (I) as claimed in claim 9, comprising hydrolyzing the compound of the formula (XV)

- 10 comprising the steps of:
  - i) adding an organic solvent to compound of formula (XV),
  - ii) adjusting the pH of the resulting solution using an acid at a temperature in the range of 10 to 40 °C,
  - iii) cooing the resulting solution rapidly to -40 to 0 ° and
- 15 iv) isolating the novel amorphous monohydrate of cefdinir of the formula (I).
  - 11. The process for the preparation of novel amorphous monohydrate of cefdinir of the formula (I) as claimed in claim 9, comprising hydrolyzing the compound of the formula (XV)

comprising the steps of:

- i) adding an organic solvent to compound of formula (XV),
- ii) cooing the resulting solution to -40 to 0 ° and
- 5 iii) adjusting the pH of the resulting solution by rapid addition of an acid at a temperature in the range of 10 to 40 °C,
  - iv) isolating the novel amorphous monohydrate of cefdinir of the formula (I).
  - 12. The process as claimed in claims 10 and 11, wherein the organic solvent is selected from acetone, 2-butanone, methanol, isopropanol, ethanol, THF, acetonitrile, DMAc, water and the like or mixtures thereof.
  - 13. The process as claimed in claims 10 and 11, wherein the acid is selected from HCl, sulfuric acid, formic acid, acetic acid or aromatic/aliphatic sulfonic acids.
  - 14. A compound of compound formula (XIV),

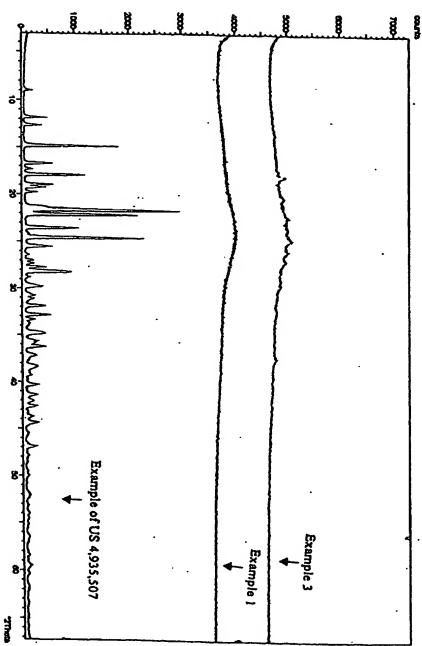
H<sub>2</sub>N 
$$\stackrel{S}{\longrightarrow}$$
  $\stackrel{O}{\longrightarrow}$   $\stackrel{Ph}{\longrightarrow}$   $\stackrel{Ph}{\longrightarrow}$   $\stackrel{COO}{\longrightarrow}$   $\stackrel{M}{\longrightarrow}$  (XIV)

wherein M<sup>+</sup> represents a counter ion.

15

WO 2004/046154 PCT/IB2003/005032





## INTERNATIONAL SEARCH REPORT

PCT/IB 03/05032

IPC 7	C07D501/06 C07D501/22	•	
According to	International Patent Classification (IPC) or to both rational classifica	Son and IPC	
B. RELDSS	EARCHED  amontation searched (classification system tollowed by classification	o contrib)	
IPC 7		u syusus)	
	·	· <u> </u>	
Documentation	on searched other their unintrium documentation to the extent that su	sch documents are included in the fields se	erched
		·	
	to been consulted during the international search (name of data base		
EPO-INT	ernal, WPI Data, PAJ, BEILSTEIN Dat	a, CHEM ABS Data	
C. DOCUME	HTS CONSIDERED TO BE RELEVANT		
Category*	Cliation of document, with indication, where appropriate, of the rela	want pessages	Relevant to claim No.
X	WO 01 79211 A (OTSUKA KAGAKU KK) 25 October 2001 (2001-10-25)	ł	1,9,14
	claims; example 1		
Y	WO 99 51607 A (DECRISTOFORO MARTI	N.	1 <b>-8</b>
'	;BIOCHEMIE GMBH (AT); STURM HUBER	T_(AT);	
	LUDES) 14 October 1999 (1999-10-1 page 1; claims; examples	4)	
l I			
Y	US 6 388 070 B1 (DESHPANDE PANDUR BALWANT ET AL) 14 May 2002 (2002		1-8
	cited in the application	. 03 247	
1	claims; examples		
P,Y	HO D2 098884 A (CHANG YOUNG KIL;		1-8
<b>!</b> !	KYUNS (KR); KIM HONS SUN (KR); LĒ 12 December 2002 (2002—12—12)	E GWAN)	
	claims; examples		
1	<del></del>	·/	
N 50	er documents are listed in the continuation of box C.	Y Patent family members are listed	h annow
		Y Petent family members are listed	
	legaries of clied documents : mi defining the general state of the ext which is not	"I" taler document published after the trite or priorby date and not be conflict with	the application but
consid	heed to be of particular relevance	clied to understand the principle or the invention "X" document of particular relevances the c	
Carp d		causes po configured novel or causes A, decreased or basicists insertained the c	be considered to
which disclose	is clied to existifish the publication date of exother n or other apacial reason (as apacifish)	"Y" document of particular relevance; the o cannot be considered to involve on in	rentive step when the
others		document is continued with one or mo ments, such continuation being obvior to the art.	
tester 0	int published prior to the international filling date but ian the priority date claimed	.g., quorament wemper of the same battan	temby
Date of the	actual completion of the international assects	Date of realing of the international so	erch report
1	9 February 2004	01/03/2004	
Name and r	neding address of the ISA European Patera Office, P.B. 5818 Patentinan 2	Authorized officer	

## INTERNATIONAL SEARCH REPORT

ptication No PCT/IB 03/05032

Clication of document, with indication, where appropriate, of the intervent passages	Relevant to claim No.
WO 03 050124 A (KUMAR NEELA PRAVEEN ;KUMAR YATENDRA (IN); PRASAD ASHOK (IN); PRASA) 19 June 2003 (2003-06-19) claims; examples	1-8,14
, , , , , , , , , , , , , , , , , , ,	
	·
į.	
	WO 03 050124 A (KUMAR NEELA PRAVEEN ;KUMAR YATENDRA (IN); PRASAD ASHOK (IN); PRASA) 19 June 2003 (2003-06-19) claims; examples

## INTERNATIONAL SEARCH REPORT

intermetten on peters tempy members

procession No PCT/IB 03/05032

Patent document cited in search report		Publication date		Patent family member(s)		Publication date
WO 0179211	A	25-10-2001	JP	2001294590 #	1	23-10-2001
			CN	1134445 8	3	14-01-2004
			EP	1273587 A	<b>41</b>	08-01-2003
		• •	MO	0179211 #	<b>41</b>	25-10-2001
NO 9951607 A 14-10-19	14-10-1999	AT	406773 8	3	25-08-2000	
			AT	57598 <i>f</i>	A	15-01-2000
			AU	3603599 /	A	25-10-1999
			BR	9909898	A	26-12-2000
			CA	. 2326441 /	<b>A1</b>	14-10-1999
•			CN	1134446 E	В	14-01-2004
			WO	9951607	A2	14-10-1 <del>99</del> 9
			EP	1068211	A2 .	17-01-2001
			ID	26210 /	A `	07-12-2000
			JP	2002510694	T	09-04-2002
			TR	200002838	<b>T2</b>	21-02-2001
		US	2003208065	A1	06-11-2003	
			ZA	200004899	A	14-10-2001
US 6388070 B1 14-05-2002	14-05-2002	CA	2433783	A1	11-07-2002	
		EP	1347970	A1	01-10-2003	
			WO	02053563	Al	11-07-2002
WO 02098884	A	12-12-2002	KR	2002092612	A	12-12-2002
			MO	02098884	A1	12-12-2002
WO 03050124	A	19-06-2003	MO	03050124	A1	19-06-2003